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Acids, bases and alkalis are found in the laboratory and at home. Acids and bases can neutralise each other. A base that can dissolve in water is also called an alkali.

*Acids and bases test questions - KS3 Chemistry Revision ...*

Indicators are used to determine whether a solution is acidic or alkaline. Acids react with metals, bases and carbonates to produce salts. Neutralisation is the reaction between an acid and a base.

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a. Acid Base Practice Test Answer Section MULTIPLE CHOICE 1. ANS: B DIF: I OBJ: 15-1.1 2. ANS: C DIF: I OBJ: 15-1.1 3. ANS: A DIF: I OBJ: 15-1.1

*Acid Base Practice Test*

Acid and Base Worksheet - Answers. 1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a)  $\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$ .  $\text{HNO}_3$  and  $\text{NO}_3^-$  make one pair  $\text{OH}^-$  and  $\text{H}_2\text{O}$  make the other. b)  $\text{CH}_3\text{NH}_2 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{NH}_3^+ + \text{OH}^-$

*Acid and Base Worksheet - Answers*

Bases turn litmus paper red. 9. Litmus paper that does not change color would indicate a pH of 7. 10. Acids would have OH in their chemical formula. 11. When you combine an acid and a base a salt and water are formed. 12. A neutralization reaction is a single replacement reaction.

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The salts of strong acids and weak bases give acidic solution having pH less than 7. Example,  $\text{NH}_4\text{Cl}$ , Ammonium Chloride will have pH less than 7. The salts of weak acids and strong bases give basic solution having pH more than 7. Example,  $\text{Na}_2\text{CO}_3$ , Sodium Carbonate will have pH more than 7. Question 7.

## *Acids, Bases and Salts Class 10 Important Questions with ...*

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Play this game to review Acids & Bases. A solution has a pH of 10. This solution can be described as \_\_\_\_\_.

## *Acids and Bases Unit Test Practice Quiz - Quizizz*

A strong acid or base is 100% ionized in aqueous solution; a weak acid or base is less than 100% ionized. The overall reaction progress stops because the reverse process balances out the forward process. pH is a measure of the hydrogen ion concentration.

## *10.E: Acids and Bases (Exercises) - Chemistry LibreTexts*

Class 7 Science Chapter 5 MCQ (Multiple Choice Questions) of

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Acids, Bases and Salts. All the important questions from chapter 5 of grade 7 science are given here in the form of MCQs for the preparation of class assessments, school tests and terminal exams.

## *Class 7 Science Chapter 5 MCQ of Acids, Bases and Salts ...*

There are several sets of definitions used to distinguish between acids and bases, but Boyle's method is easy and practical. Acids tend to be sour-tasting (don't taste them!), they are corrosive, and they turn litmus paper red. Bases are slippery-feeling, soapy-tasting, and turn litmus paper blue.

## *Acids and Bases Chemistry Quiz - ThoughtCo*

1. Both acids and bases: A have a sour taste B contain hydrogen ions in solution C are corrosive: 2. The pH of acids is: A greater than 7 B equal to 7 C less than 7: 3. Bases turn litmus paper: A blue B purple C red: 4. The name given to a soluble base is: A an acid B a dissolvable base C an alkali: 5. Vinegar, aspirin and vitamin C are examples of household:

## *Acids and Bases Quiz - Qld Science Teachers*

Tri acidic bases are those which have three 'OH' group per unit as  $\text{Al}(\text{OH})_3$  and  $\text{Fe}(\text{OH})_3$ , their one mole requires three mole of a base (NaOH) for complete neutralization. 365 366 367 Answer

## *Answers about Acids and Bases*

Turmeric is a natural indicator. On adding its paste to acid and base separately, which colours would be observed. (a) Yellow in both acid and base (b) Yellow in acid and red in base (c) Pink in acid and yellow in base (d) Red in acid and blue in base. Answer. Answer: (b) Yellow in acid and red in base

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