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Worksheet. 1)

Explain how you  
would make 450

mL of a 0.250 M

NaOH solution. Add  
water to 4.52

grams of sodium  
hydroxide until the  
final volume of the  
solution is 450 mL.

2) To what volume  
will you have to  
dilute 30.0 mL of a

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12 M HCl solution  
to make a 0.35 M  
HCl solution?

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Worksheet.doc -

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Solutions 1) Explain

how you would

make 750. mL of a

1.35 M KOH

solution. 0.750 L

KOH x 1.35 mole

KOH x 59 g KOH =

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56.7 g KOH 1 L  
KOH 1 mole KOH  
Measure 56.7 g  
KOH into a  
containers and add  
water to 750 mL  
volume. 2) If you  
dilute 15.0 mL of a  
12.0 M HCl solution  
to make a 2.50 M  
HCl solution what  
will the final  
volume be?

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Worksheet -  
Everett Community  
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Worksheet. 1)

Explain how you  
would make 450  
mL of a 0.250 M  
NaOH solution. Add  
water to 4.52  
grams of sodium  
hydroxide until the



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final volume of the solution is 450 mL.  
2) To what volume will you have to dilute 30.0 mL of a 12 M HCl solution to make a 0.35 M HCl solution? 1030 mL

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1) Explain how you  
would make 450  
mL of a 0.250 M  
NaOH solution. Add  
water to 4.52  
grams of sodium

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2) To what volume will you have to dilute 30.0 mL of a 12 M HCl solution to make a 0.35 M HCl solution? 1030 mL

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1) Explain how you would make 450 mL of a 0.250 M NaOH solution.

Add water to 4.52 grams of sodium hydroxide until the final volume of the solution is 450 mL.

2) To what volume will you have to dilute 30.0 mL of a 12

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the end of your  
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#### Making Solutions Worksheet -

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Making a Solution  
How to make 1L of  
a 5M solution of a  
substance with a  
molecular weight of  
75 g/mol. How  
many grams of the

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Solute should we  
weight out?  
Approach: figure  
out how many  
moles we need,  
then convert to  
grams.  $CV = \text{Total}$   
amount  $5 \text{ mol/L} \times 1$   
 $L = 5 \text{ mol}$  Grams =  
moles  $\times$  grams/mol  
(MW)  $5 \text{ mol} \times 75$   
 $\text{g/mol} = 375 \text{ g}$

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Solutions and  
Dilutions  
Practice Work  
Chemistry

Solutions Practice  
Problems 1. Molar  
solutions. a.

Describe how you  
would prepare 1 L  
of a 1 M solution of  
sodium chloride.

The gram formula  
weight of sodium  
chloride is 58.44  
g/mol. Answer: To



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make a 1 M solution of sodium chloride, dissolve 58.44 g sodium chloride in 500 mL water in a 1000-mL volumetric flask. When all the solid is dissolved and the solution is at room temperature, dilute to the mark and invert the flask several times to

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## Practice Work Chemistry Solutions Practice Problems |

Carolina.com

First, express the percent of solute as a decimal:  $5\% = 0.05$ . Multiply this decimal by the total volume:  $0.05 \times 1000\text{ml} = 50\text{ml}$  (ethylene glycol

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Answers

needed). Subtract the volume of solute (ethylene glycol) from the total solution volume: 1000ml (total solution volume) - 50ml (ethylene glycol volume) = 950ml (water needed)

Preparing Chemical  
Solutions - The

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Solutions Company

Example 1: To prepare a liter of a molar solution from a dry reagent .

Multiply the molecular weight (or FW) by the desired molarity to determine how many grams of reagent to use:

Suppose a compound's MW =

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194.3 g/mole; to  
make 0.15 M  
solution use 194.3  
g/mole \* 0.15  
moles/L = 29.145  
g/L

Resource Materials:  
Making Simple  
Solutions and  
Dilutions

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practice, print a  
copy of this quiz at

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the Chemistry:

Solutions webquest  
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Solutions –

Ducksters

Practice Work  
Then, work through  
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Answers

your whole  
process, and show  
the actions and  
decisions in the  
order that they  
happen. Link them  
with arrows to  
illustrate the flow  
of the process.

Where you need to  
make a decision,

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draw arrows from the decision diamond to each possible solution, and then label each arrow with the decision made.

Flow Charts -  
Problem-Solving  
Skills From  
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Practice: Solutions  
and mixtures.



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Practice:

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Solutions and mixtures. Up Next. Solutions and mixtures. Our mission is to

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provide a free,  
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anyone, anywhere.

## Answers

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calculations  
(practice) | Khan  
Academy

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corresponding  
worksheet will

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Show And  
Answers

gauge your understanding of solutions in chemistry. Topics you'll need to know to pass the quiz include solutions and their ...

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Chemistry |  
Study.com  
Answer to Show

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the calculations for making the 20mL of 0.2M, 0.1M, and 0.05M  $\text{Cu}^{2+}$  solutions. I am not sure how to do this. Please ...

Solved: Show The Calculations For Making The 20mL Of 0.2M ...

In addition to the above food for

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Answers

thought, consider trying out the following strategies to improve your no-show rates. 1.

Make Daily

Reminder Calls.

Designate a staff member to make reminder calls each afternoon for the next day's appointments.

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The Best Ways to  
Reduce Patient No-  
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