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Key Exercise 12.1 - Using Electronegativities: Classify the following bonds as nonpolar covalent, polar covalent, or ionic. If a bond is polar covalent, identify which atom has the partial negative charge and which has the partial positive charge. If a bond is ionic, identify which atom

Chapter 12 Liquids: Condensation, Evaporation, and Dynamic ...

Chemistry Chapter 12 Study Guide. calorimetry. heat capacity. specific heat. thermochemistry. the study of heat flow and measurement. the amount of heat needed to raise the temperature of an object. heat capacity of 1 gram of a substance. the study of the changes in heat in chemical reactions.

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ch 12 Study guide TE - Mr. McKnight Clawson High School 240 Chemistry: Matter and Change □ Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

## States of Matter

### CHAPTER STUDY GUIDE FOR CONTENT MASTERY

Section 12.2 Stoichiometric Calculations In your textbook, read about mole-to-mole conversion. Read the following passage and then solve the problems. In the equation that follows each problem, write in the space provided the mole ratio that can be used to solve the problem.

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This Study Guide for Content Mastery for Chemistry: Matter and Change will help you learn more easily from your textbook. Each textbook chapter has six study guide pages of questions and exercises for you to complete as you read the text. The study guide pages are divided into sections that match those in your text.

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